

(Pure substances and mixtures)

Q1. (2022)

$$\text{mass} = \text{density} \times \text{volume}$$

$$\text{mass} = 0.40 \times 40$$

$$= 16 \text{ (g)}$$

Q2.(2023)

$$(M_r = 63.5 + 56 + (2 \times 32) =) \\ 183.5$$

$$(\% \text{ of copper} =) \frac{63.5}{183.5} \times 100$$

$$= 34.6 \text{ (\%)}$$