

Group 0 – Noble gases

A1. (2019)

03.3	noble gases		1	AO1 5.1.2.4
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A2. (2019)

03.7	stable arrangement (of electrons)	allow full outer shell allow eight electrons in the outer shell allow does not need to gain or lose electrons	1	AO1 5.1.2.4
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A3. (2022)

04.4	any one from: • argon has a full outer shell • argon has 8 electrons in the outer shell • argon has a stable arrangement of electrons.	allow energy level for shell	1	AO2
	(so) argon is unreactive	allow does not need to lose and / or gain electrons ignore argon is a noble gas	1	AO1 5.1.2.4

A4. (2023)

03.1	noble gases		1	AO1 5.2.1.2
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A5. (2023)

05.4	argon		1	AO3 5.1.2.4
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