

Soluble salts (practical preparation)

Q1. (2018)

03	Level 3: Relevant points (reasons / causes) are identified, given in detail and logically linked to form a clear account.	5–6	AO3
	Level 2: Relevant points (reasons / causes) are identified, and there are attempts at logical linking. The resulting account is not fully clear.	3–4	AO3
	Level 1: Points are identified and stated simply, but their relevance is not clear and there is no attempt at logical linking.	1–2	AO1
	No relevant content	0	
	Indicative content: <ul style="list-style-type: none"> uses sulfuric acid not hydrochloric acid or sulfuric acid needed uses copper carbonate / oxide not calcium carbonate or copper carbonate / oxide needed add solid until solid remains or is in excess or no more reacts / dissolves so that most / all of the acid reacts filter to remove excess or unreacted carbonate / oxide / solid heat gently or partially evaporate or leave until crystals appear or to crystallise <p>for level 3 the correct chemicals must have been selected</p>		5.4.2.3

Q2. (2020)

04.2	any one from: <ul style="list-style-type: none"> warm / heat the mixture increase the concentration of the (hydrochloric) acid 	ignore add a catalyst ignore stir ignore powder ignore add more zinc oxide do not accept volume / amount of (hydrochloric) acid do not accept increase the surface area	1	AO1 5.4.2.2 5.4.2.3 RPA8
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Q3. (2020)

04.3	zinc oxide remains or solid remains	ignore colour allow zinc oxide is added until in excess	1	AO1 5.3.2.4 5.4.2.2 5.4.2.3 RPA8
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Q4. (2020)

04.4	filtration / filter		1	AO1 5.4.2.2 5.4.2.3 RPA8
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Q5. (2020)

04.5	heat	do not accept heat to dryness	1	AO1 5.4.2.2 5.4.2.3 RPA8
	leave to crystallise / cool	allow leave to evaporate some water	1	

Q6. (2021)

02.6	Level 3: The method would lead to the production of a valid outcome. The key steps are identified and logically sequenced.		5–6	AO1 5.4.2.2 5.4.2.3 RPA8
	Level 2: The method would not necessarily lead to a valid outcome. Most steps are identified, but the method is not fully logically sequenced.		3–4	
	Level 1: The method would not lead to a valid outcome. Some relevant steps are identified, but links are not made clear.		1–2	
	No relevant content		0	
	Indicative Content: <ul style="list-style-type: none"> • react hydrochloric acid • (with) copper oxide • in a suitable container • warm (hydrochloric) acid • add copper oxide • until is in excess or until solid remains • stir • filter excess copper oxide • pour solution / filtrate into evaporating basin • use of water bath or use of electric heater • to heat gently or partially evaporate • leave to cool / crystallise For level 3 the correct chemicals must have been selected.			

Q7. (2022)

01.2	filtration or filter		1	AO1 5.4.2.3 RPA8
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Q8. (2024)

02.3	heat the solution (to remove some of the water)	ignore references to filtration allow evaporate some of the water / solution	1	AO1 5.1.1.2 5.4.2.3 RPA8
	leave to cool / crystallise		1	